

# THE PROBLEM OF NON-CONDENSABLE GAS RELEASE IN EVAPORATORS

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## Contents

1. Introduction
  2. Sources of Non-condensable Gases in Evaporators
  3. Effects of Non-condensable Gases in Evaporators
  4. Summary of Cases that Require Consideration of Non-condensable Gas Release
- Bibliography and Suggestions for further study

## Summary

The presence of non-condensable gases in desalination distillers, particularly carbon dioxide, nitrogen, oxygen, and argon, is caused by air leakages and the release of dissolved gases from the evaporating brine. The gases which are molecularly dissolved in seawater, namely nitrogen, oxygen, and argon, can be removed almost completely in a deaerator. When no deaerator is provided, they will be released from the evaporating brine and must be extracted by adequate venting. Carbon dioxide, which reacts chemically in seawater, cannot be removed by simple deaeration. For the removal of carbon dioxide, the addition of a strong acid to the feedwater and a decarbonator are required. Without decarbonation, carbon dioxide is released from the brine during evaporation and has to be extracted by adequate venting. The presence of non-condensable gases affects the heat transfer for condensation, energy consumption, operation, and material lifetime of desalination distillers.

## 1. Introduction

As in many other industrial distillation processes, the presence of gases which do not condense at the operating conditions of the process, in the following designated as "non-condensable gases", is a serious problem in seawater distillation. Since removal of non-condensable gases is vital to the efficient operation of all desalination distillers, each plant is equipped with a venting system. Deaerators and decarbonators are used in certain cases to remove the gases from the feedwater before they enter the evaporator.

The knowledge of which gases are present and in what quantities, how the quantities are distributed between the individual distiller stages, and how the gases affect the performance, energy consumption, and material lifetime of the distiller is of great importance in the design, operation, and costs of a desalination plant.

In the following, the sources of non-condensable gases, the effects of the gases in

desalination distillers, and the cases that require consideration of the gas release are summarized.

## 2. The Sources of Non-condensable Gases in Evaporators

The presence of non-condensable (NC) gases in seawater evaporators is caused by the leakage of ambient air through flanges, man-holes, instrumentation nozzles, etc., into the parts of the evaporator operating under vacuum (Gregorzewski et al. 1993) and the release of dissolved gases from the evaporating brine.

All the gases present in the atmosphere dissolve in seawater to a greater or lesser extent. As far as quantity is concerned, only four gases are of interest for engineering applications, namely nitrogen ( $N_2$ ), oxygen ( $O_2$ ), argon (Ar), and carbon dioxide ( $CO_2$ ). Notwithstanding its low concentration in the air, it is mainly  $CO_2$  which determines the nature of seawater to an outstanding degree. This arises from the fact that  $CO_2$  is involved in the carbonate system of seawater which regulates the pH of the seawater.

The released NC gases can be classified into the following.

- (a) Those which are molecularly dissolved in seawater and do not react chemically such as  $N_2$ ,  $O_2$ , and Ar.
- (b) Those which react chemically in seawater and are produced during the desalination process itself such as  $CO_2$ . Under the alkaline conditions prevailing in seawater, only a small proportion of the total inorganic carbon content in seawater is present as molecular  $CO_2$  gas and the major proportion is chemically combined in bicarbonate  $HCO_3^-$  and carbonate  $CO_3^{2-}$  ions. During the desalination process  $CO_2$  is released from the evaporating brine. Thus, the equilibria between  $HCO_3^-$ ,  $CO_3^{2-}$ , and  $CO_2$  in the brine are disturbed and new molecular  $CO_2$  is produced by chemical reactions.

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